

Assignment 8-3, 8-4, 8-5

Name _____

Class Period _____

Date _____

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1. What is a **synthesis reaction**?

2. Write the equation for a **synthesis reaction** on the line below.

3. What is a **decomposition reaction**?

4. Write the equation for a **decomposition reaction** on the line below.

5. What is a **single-replacement reaction**?

6. Write the equation for a **single-replacement reaction** on the line below.

7. What is a **double-replacement reaction**?

8. Write the equation for a **double-replacement reaction** on the line below.

9. Who discovered the element **oxygen**?

10. Describe the chemical reaction that led to the discovery of oxygen.

11. What type of reaction occurs when baking soda is mixed with vinegar?

12. Write the chemical formulas for baking soda and vinegar.

13. Describe the chemical reaction that allows us to get silver metal from a solution of silver nitrate.

Please continue on the other side.

14. What does the prefix **exo** mean?

15. What does the prefix **endo** mean?

16. Explain the difference between an exothermic reaction and an endothermic reaction.

17. Give two examples of chemical reactions that require extra **activation energy** to get started. (More than the energy available from a room temperature atmosphere.)

18. What is meant by the term **reaction rate**?

19. Basing your answer on the *collision theory*, explain why the **concentration** of reactants affects the reaction rate.

20. Basing your answer on the *collision theory*, explain why the **amount of surface area** of reactants affects the reaction rate.

21. Basing your answer on the *collision theory*, explain why the **temperature** of reactants affects the reaction rate.

22. Explain how **catalysts** increase the reaction rate of chemical reactions.

23. What is the **greenhouse effect**?

24. How does carbon dioxide (CO₂) contribute to excess warming of our atmosphere?

25. According to your science text, how much carbon dioxide is released into the atmosphere every year?
