Physical Science Assignment 3-1	Name Class Period	
Read section 3-1. Then, <b>based on the information</b> <b>in your textbook, answer the following questions :</b> 1. <b>physical properties</b>		
List the four <i>phases</i> of matter?		
What is a <i>crystal</i> ?		
What are <b>amorphous solids</b> ?	<u> </u>	
List four examples of <i>amorphous solids</i> .		
What is <i>viscosity</i> ?		
Give an example of a liquid that has a <b>higher</b> viscosity than a Explain why the basic characteristic of all gases makes it ver planet (like Mercury) to have an atmosphere.		
What does Boyle's Law say?		
0. In the boxes below, draw diagrams that show the motion an <b>solid</b> , <b>liquid</b> and <b>gas</b> phase.	d spacing of particles in the	

liquid

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solid

- 11. Explain why a liquid or a gas can take the shape of its container, but a solid cannot.
- 12. How many collisions per second does a gas particle undergo?
- 13. What causes the outward pressure on a container of gas?
- 14. If we squeeze 2 liters of gas into a 1 liter space, what happens to the interior gas pressure on the walls of the container?
- 15. If we allow 2 liters of gas to expand into a 4 liter container, what happens to the interior gas pressure on the walls of the container?
- 16. If we have 1 liter of gas in a sealed balloon at room temperature then heat the balloon and gas another 10°C, what will happen to the volume of the gas?
- 17. If we have 1 liter of gas in a sealed balloon at room temperature then cool the balloon and gas to 10°C below room temperature, what will happen to the volume of the gas?
- 18. Describe the fourth phase of matter (plasma).
- 19. Where is most plasma located?
- 20. Describe how glass is formed.
- 21. In the chart below, rank the following liquids in order of highest viscosity to lowest viscosity with **10 being the highest** and **3 being the lowest**.

Name of Liquid	Viscosity Ranking
rubbing alcohol	
water	
liquid dish soap	
completely melted ice cream	
roofing tar	
ranch style salad dressing	
vegetable oil	
honey	